CHAPTER - AMINES

MULTIPLE CHOICE QUESTIONS

TOPIC- Preparation Of Amines

- **Q.1** Which of the following compound will give secondary amine on the reduction?
- a) Alkyl nitrile b) Carbylamine c) Primary amine d) Secondary nitro compound

Answer. (b) Carbylamine

- **Q.2** In order to prepare a 1° amine from an alkyl halide with simultaneous addition of one CH₂ group in the carbon chain, the reagent used as source of nitrogen is:
- (a) Sodium amide, NaNH₂
- (b) Sodium azide, NaN₃
- (c) Potassium cyanide, KCN
- (d) Potassium phthalimide, C₆H₄(CO)₂N⁻K⁺

Answer:(c) Potassium cyanide, KCN

- **Q.3** Which of the following reagents would not be a good choice for reducing an aryl nitro compound to an amine?
- (a) H₂ (excess)/Pt
- (b) LiAlH₄ in ether (c) Fe and HCl
- (d) Sn and HCl

Answer:(b) LiAlH₄ in ether

- Q.4 The best reagent for converting 2-phenylpropanamide into 2-phenylpropanamine is:
- (a) excess H₂

- (b) Br₂ in aqueous NaOH
- (c) iodine in the presence of red phosphorus
- (d) LiAlH₄ in ether

Answer:(d) LiAlH4 in ether

- **Q.5** Amongst the given set of reactants, the most appropriate for preparing 2° amine is:
- (a) 2° R—Br + NH₃
- (b) 2° R—Br + NaCN followed by H₂/Pt
- (c) 1° R—NH₂ + RCHO followed by H₂/Pt
- (d) 1° R—Br (2 mol) + potassium phthalimide followed by H₃O⁺/heat

Answer:(c) 1° R—NH₂ + RCHO followed by H₂/Pt

- Q.6 Reduction of CH₃CH₂NC with hydrogen in presence of Ni or Pt as catalvst gives
- (a) CH₃CH₂NH₂

(b) CH₃CH₂NHCH₃

(c) CH₃CH₂NHCH₂CH₃

(d) $(CH_3)_3N$

Answer:(b) CH₃CH₂NHCH₃

- Q.7 The best reagent for converting, 2-phenylpropanamide into 1-phenylethanamine is:
- (a) excess H₂/Pt

(b) NaOH/Br₂

(c) NaBH₄/methanol

(d) LiAlH₄/ether

Answer:(b) NaOH/Br₂

- Q.8 Hoffmann bromamide degradation is used for the preparation of
- (a) primary amines

(b) secondary amines

(c) tertiary amines

(d) secondary aromatic amines

Answer:(a) primary amines

- **Q.9** The source of nitrogen in Gabriel synthesis of amines is:
- (a) Sodium azide, NaN₃

- (b) Sodium nitrite, NaNO₂
- (c) Potassium cyanide, KCN
- (d) Potassium phthalimide, C₆H₄(CO)₂N⁻K⁺

Answer:(d) Potassium phthalimide, C₆H₄(CO)₂N⁻K⁺

- Q.10 Best method for preparing primary amines from alkyl halides without changing the number of carbon atoms in the chain is
- (a) Hoffmann Bromamide reaction
- (b) Gabriel phthalimide synthesis
- (c) Sandmeyer reaction
- (d) Reaction with NH₃

Answer: (b) Gabriel phthalimide synthesis

- **Q.11** Which of the following methods of preparation of amines will give same number of carbon atoms in the chain of amines as in the reactant?
- (a) Reaction of nitrite with LiAlH₄.
- (b) Reaction of amide with LiAlH₄ followed by treatment with water.
- (c) Heating alkylhalide with potassium salt of phthalimide followed by hydrolysis.

(d) Treatment of amide with bromine in aqueous solution of sodium hydroxide.

Answer: (c) Heating alkylhalide with potassium salt of phthalimide followed by hydrolysis.

Q.12 Reduction of aromatic nitro compounds using Fe and HCl gives ______

(a) aromatic oxime

(b) aromatic hydrocarbon

(c) aromatic primary amine

(d) aromatic amide

Answer: (c) aromatic primary amine

Q.13 Reduction of nitrobenzene by which of the following reagent gives aniline? (a) Sn/HCl (b) Fe/HCl (c) H₂-Pd (d) all the above

Answer: (d) all the above

Assertion and Reason Type Questions

Note: In the following questions a statement of assertion followed by a statement of reason is given. Choose the correct answer out of the following choices.

- (a) If both Assertion and Reason are correct and the Reason is a correct explanation of the Assertion.
- (b) If both Assertion and Reason are correct but Reason is not a correct explanation of the Assertion.
- (c) If the Assertion is correct but Reason is incorrect.
- (d) If both the Assertion and Reason are incorrect.
- Q.1 **Assertion**: Aromatic 1° amines can be prepared by Gabriel Phthalimide Synthesis.

Reason: Aryl halides undergo nucleophilic substitution with anion formed by phthalimide.

Answer: (d)

Q.2 **Assertion**: Ammonolysis of alkyl halide is not a suitable method for the preparation of pure primary amines.

Reason: Ammonolysis of alkyl halide yields mainly secondary amines.

Answer: (c)

Q.3 **Assertion**: Hoffmann's bromamide reaction is given by primary amines.

Reason: Primary amines are more basic than secondary amines.

Answer: (c)